

Brightsite

Transforming industry

Plasmatechnology as a gamechanger in a Circular Economy

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Proud partners

Sitech Services

TNO

Maastricht University

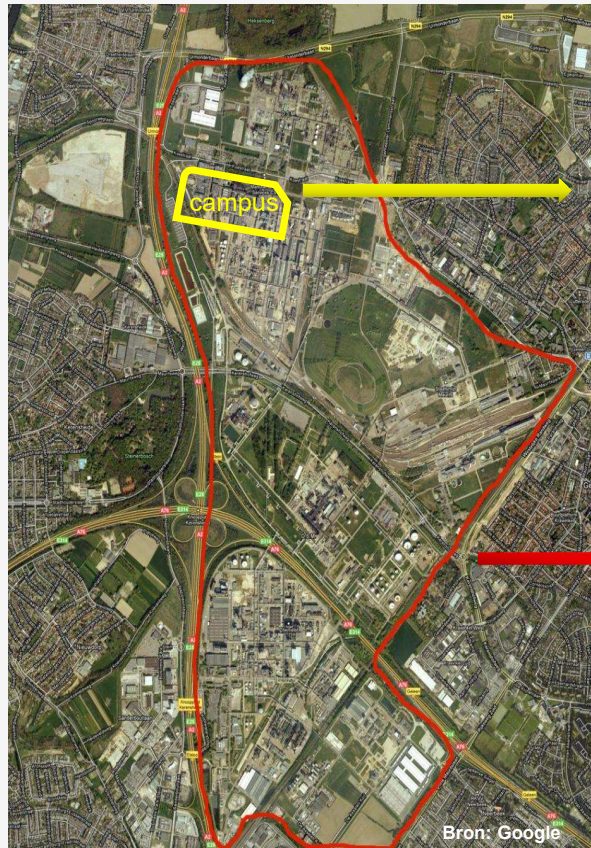
Brightlands Chemelot campus



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The chemical industry needs to drastically reduce its carbon emissions to mitigate climate change and meet sustainability goals.

The European Climate Law sets the target of reducing greenhouse gas emissions by at least 55% by 2030, compared to 1990 levels and climate neutrality by 2050.

The chemical industry is at the dawn of a revolution, drastic steps are necessary to reduce the CO₂ emission and to reach the climate goals:

- Transition to renewable energy sources: electrify what can be electrified
- Maximize CO₂ free energy generation
- Prioritize on Energy Efficiency
- **Unlock / disclose alternative feedstock (first recycling and selective biomass)**

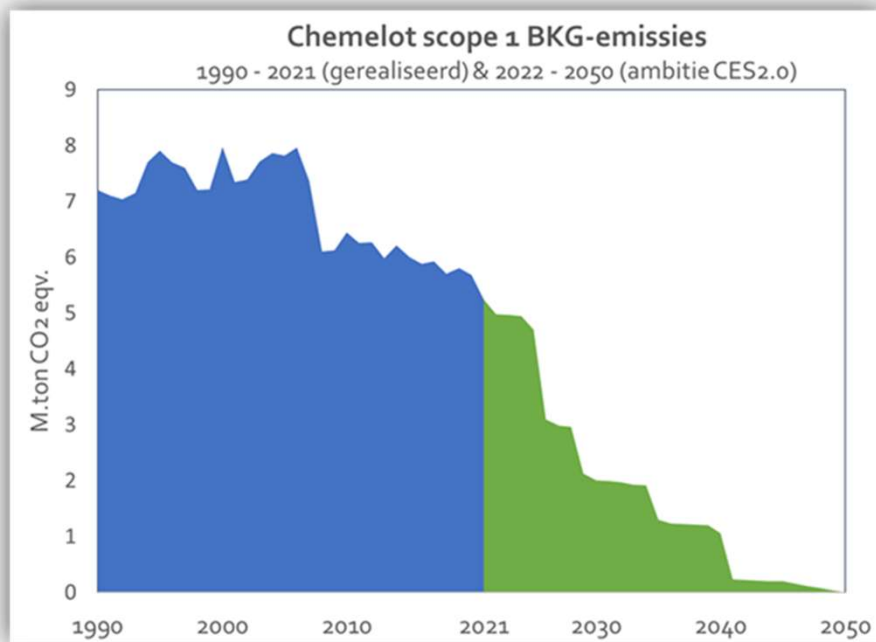


Slide 4

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Plaatje controleren

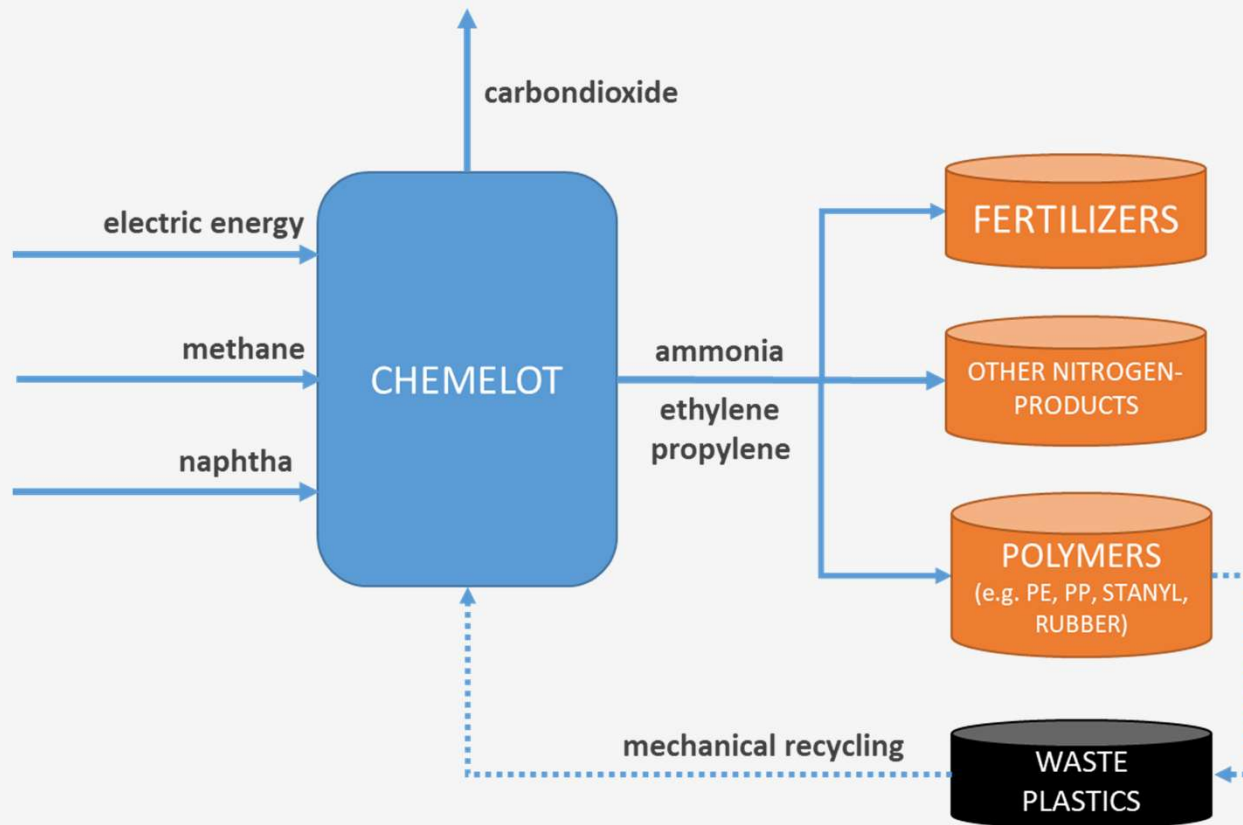
Linden, J.L. (Hans); 2023-11-15T14:11:17.219



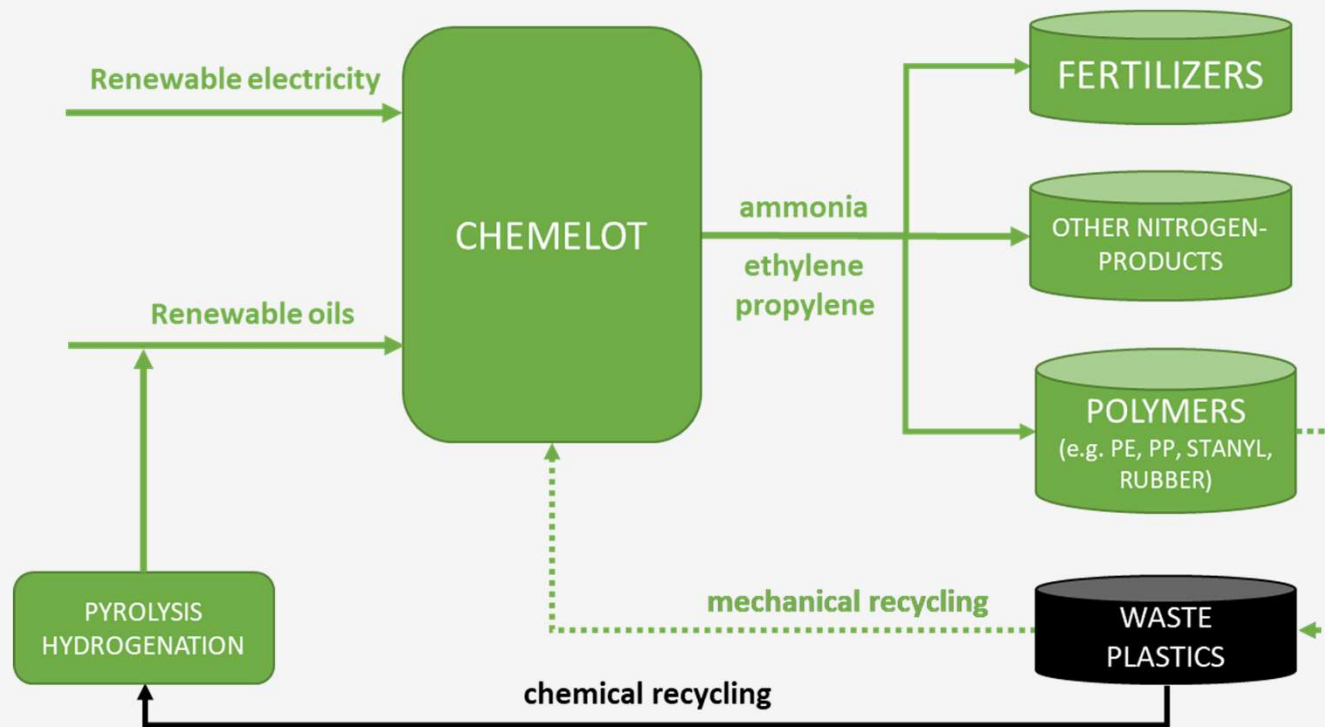
- Chemelot is one of the largest CO₂ emitters in the Netherlands
- Chemelot goal: zero emissions in 2050



Present situation Chemelot



Possible future situation Chemelot 2050



2 energy intensive routes for circular feedstock

- Synthetic molecules, e.g. H_2 and CO from H_2O and CO_2
- Renewable carbon: CO_2 , biomass and waste



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Linden, J.L. (Hans); 2023-11-15T14:11:17.219

We believe that plasma technology can play a role in helping the chemical industry to reduce its carbon emissions to mitigate climate change and meet sustainability goals

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- Fourth state of matter
- Ionized gas with equal numbers of positively charged ions and negatively charged electrons
- Electrical conductive

Examples:



Lighting



Aurora Borealis



HR+ glass coating (Pilkington)



Welding

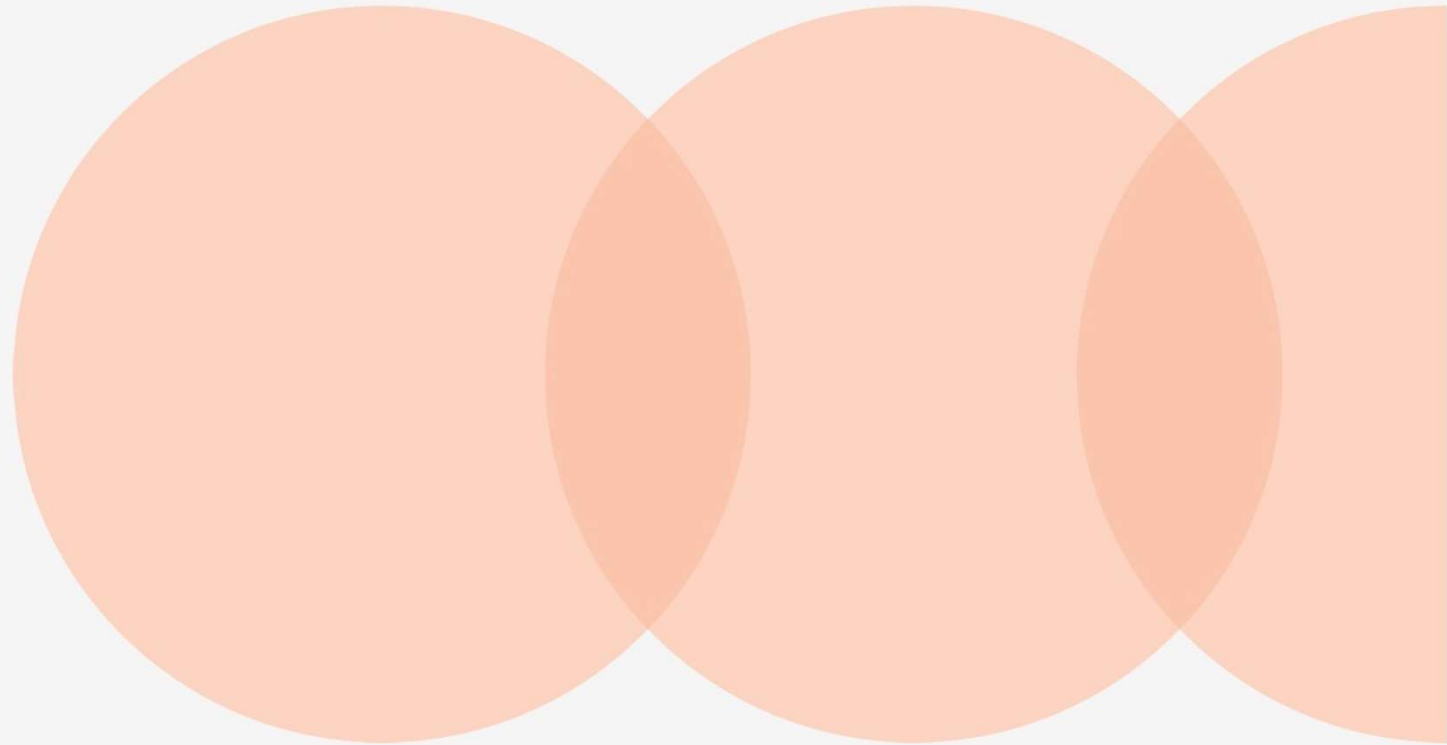


Neon Light

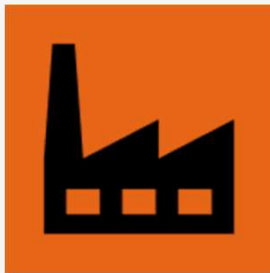
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Methane



Green house gas emission in the chemical industry

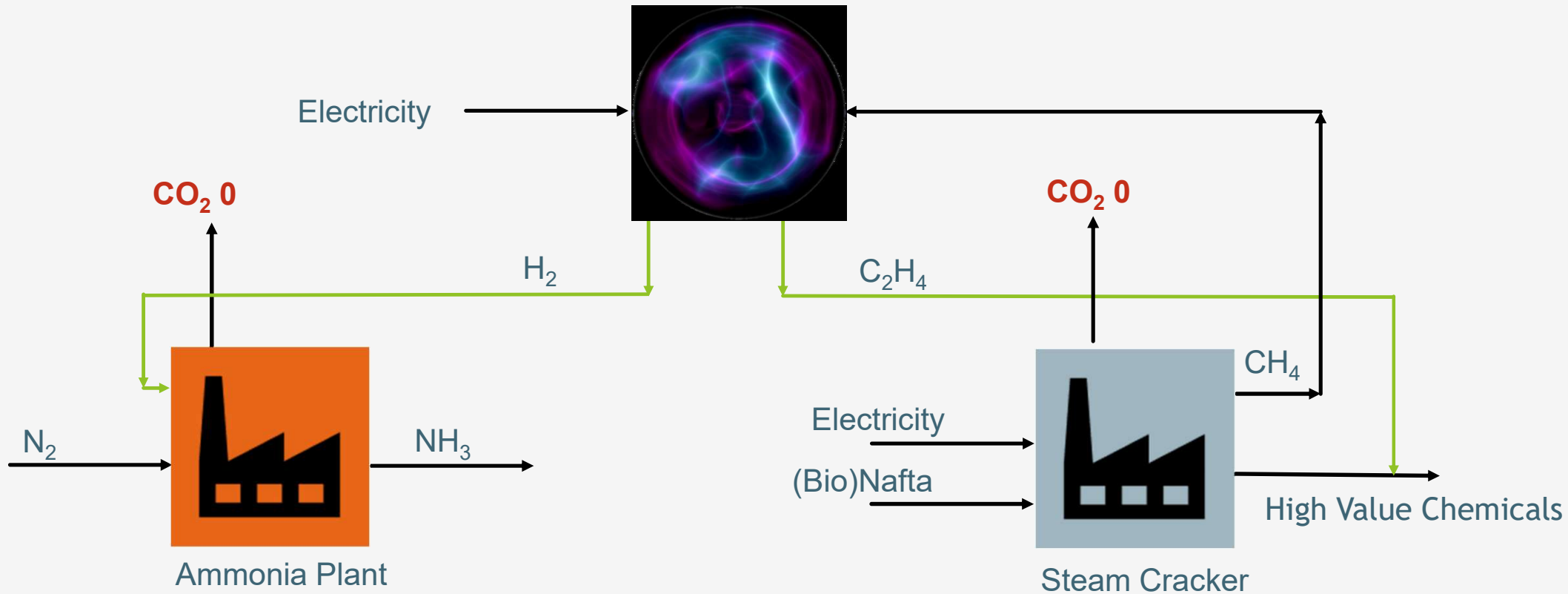


Ammonia Plant:
Typically 9 ton CO₂ is emitted
per 1 ton of hydrogen



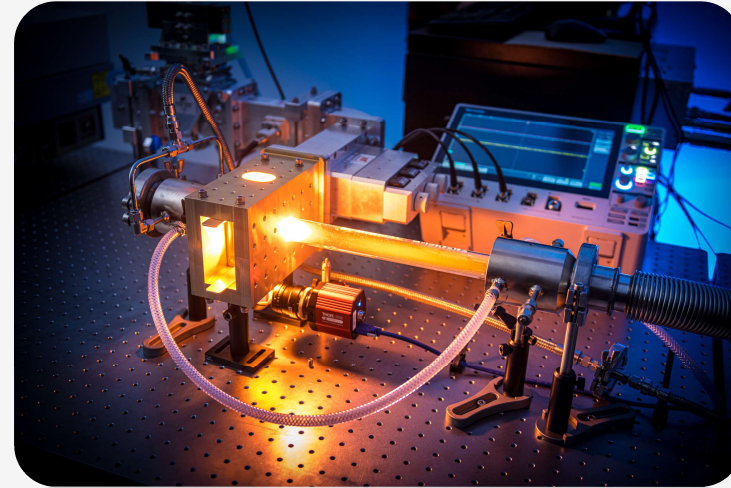
Steam Cracker:
Typically 1 ton CO₂ is
emitted per 1 ton of olefins

Combined Transition



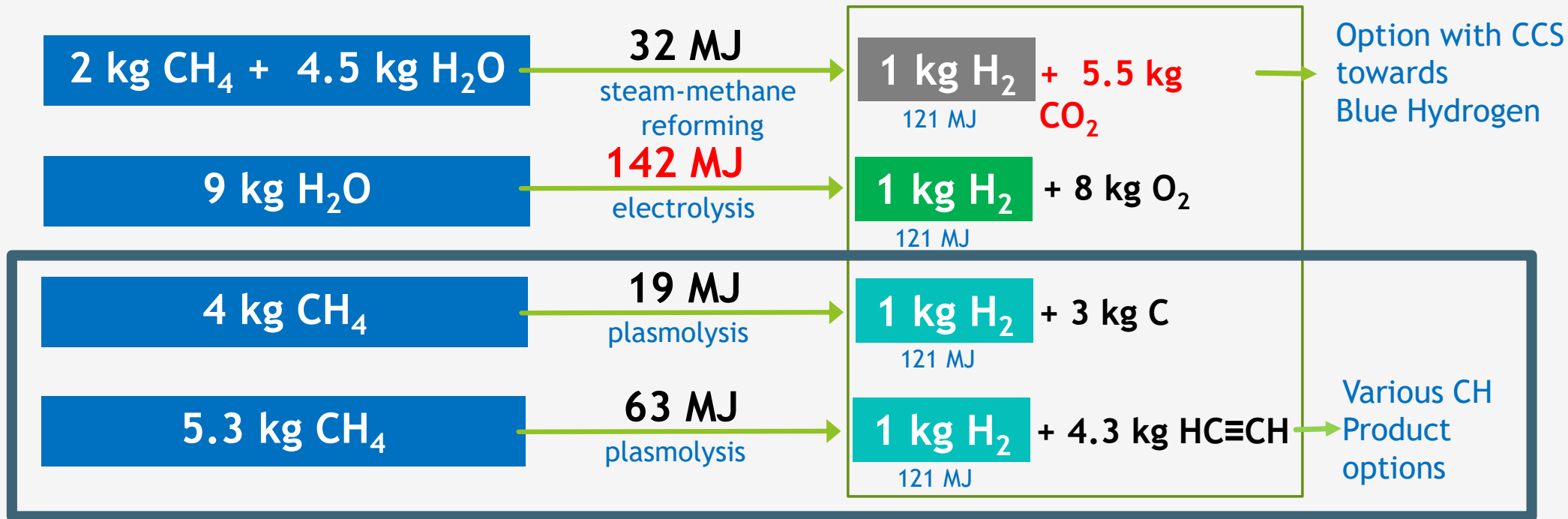
Plasma: a disruptive pyrolysis technology to transform methane into hydrogen and high-value chemicals

- Efficient production of **hydrogen and acetylene/ethylene**, building blocks to produce fertilizers and plastics
- No scope 1 CO₂ emission
- **Based on a proven technology (Hüls process)**
- Efficient heating with high temperatures
- Intermittent operation possible



Plasmas are the fourth state of matter, distinct from solids, liquids and gases. They are formed when a gas is heated to such high temperatures that its atoms or molecules become ionized, resulting in a mixture of free electrons and positively charged ions.

Motivation for plasma decarbonization



Thermodynamic numbers, no heating or cooling

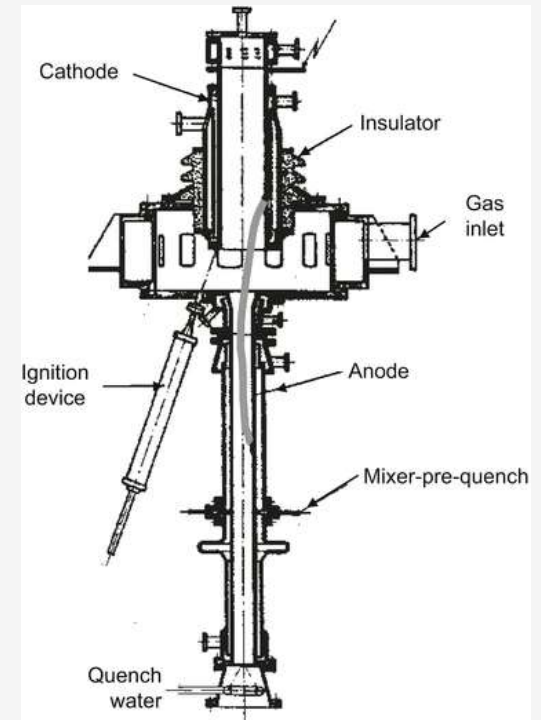
Hüls process

INSPECTION OF SYNTHETIC RUBBER PLANT OF
CHEMISCHE WERKE HÜLS, HÜLS, GERMANY

Date of Inspection: April 11, 1945

Object: This target is primarily a synthetic rubber plant using, as one of the principal materials, acetylene manufactured from natural gas. The investigation of this target was undertaken by the Oil Team to obtain the information available on the production of acetylene and other products from natural gas.

The plant consists of 14 arcs with 12 arcs running at any one time. One arc consumes 7000 kw. and produces 700-800 kgs. of acetylene per hour. The cost of manufacturing acetylene by this process is reported to be 25-30 pfennings per kg. of 95-96% acetylene. The following is the consumption of utilities per 100 kgs. of acetylene produced:



Commercial activities Monolith



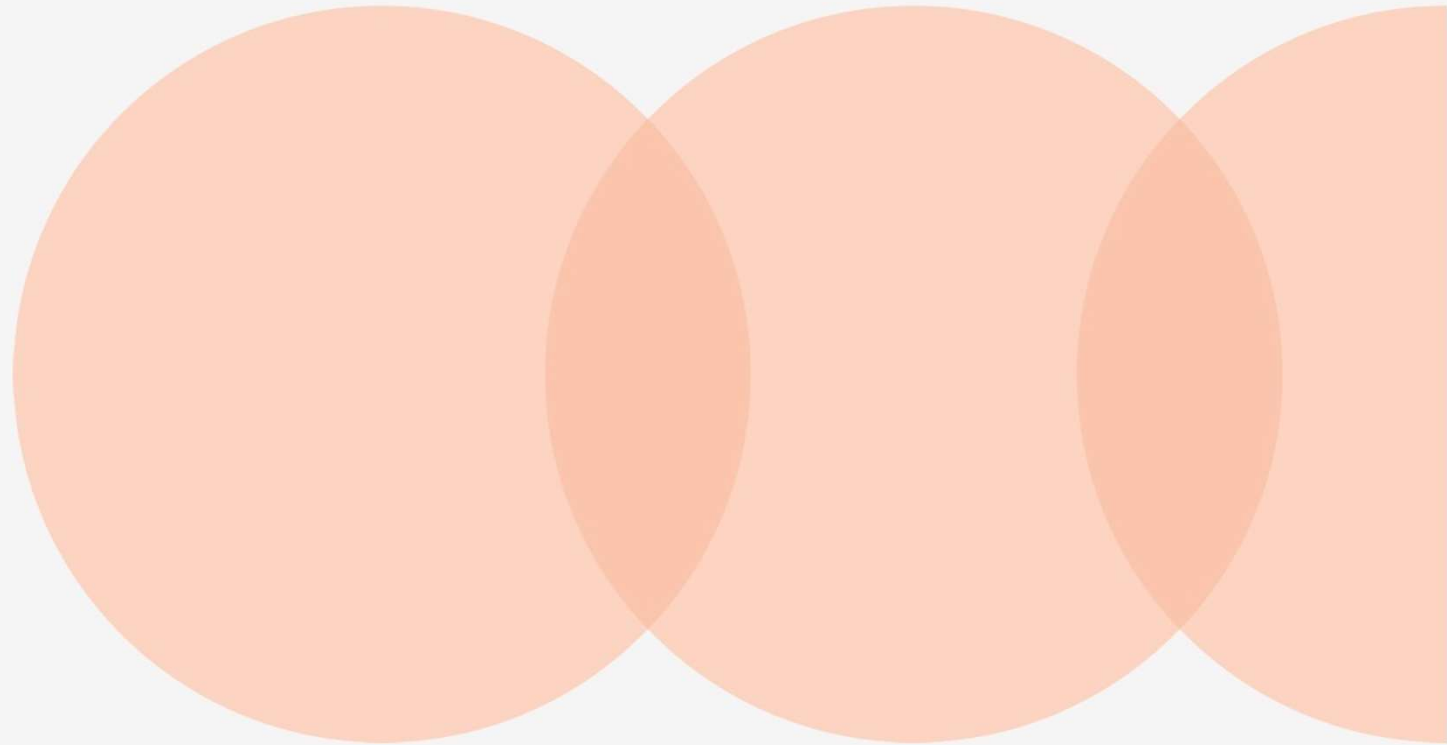
Monolith Olive Creek Expansion
Hallam, Nebraska



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Nitrogen



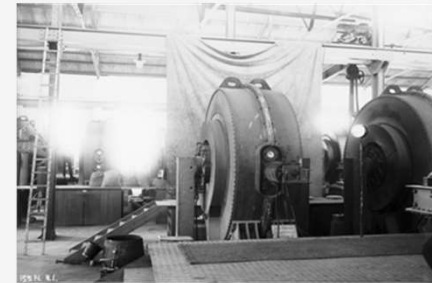
Birkeland-Eyde process

Plasma going commercial:

- Notodden, 1906 operational
- 32 arc furnaces of 750 kW
- electricity from waterfall power station, built at the same time with the factory.

Surpassed by Haber-Bosch

Source: <http://www.notoddenhistorielag.no/index.php?page=lysbueovnen>



Source: <http://no.wikipedia.org/wiki/Birkeland-Eyde-prosessen>

ON THE OXIDATION OF ATMOSPHERIC NITROGEN IN ELECTRIC ARCS.

BY KR. BIRKELAND, PROFESSOR OF PHYSICS AT THE UNIVERSITY
OF CHRISTIANIA.

(A Paper read before the Faraday Society on Monday, July 2, 1906,
PROFESSOR S. P. THOMPSON, F.R.S., in the Chair.)

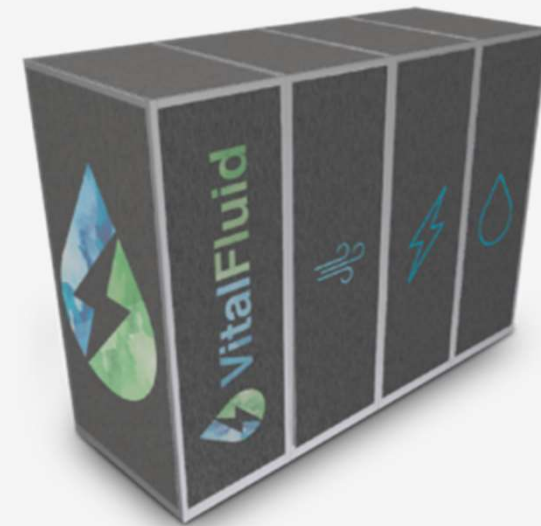
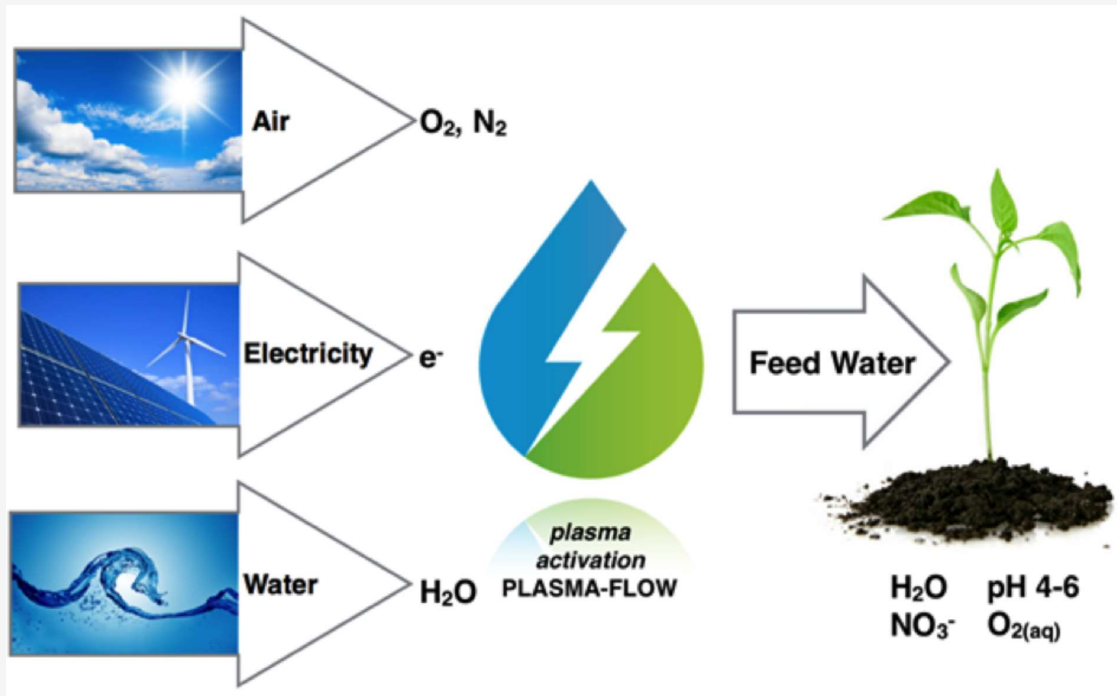
Last year, in the month of May, a factory was started at Notodden, in Norway, for the manufacture of calcium nitrate from air and limestone by the aid of electric flames (see Figs. 1 and 1a).

Thanks to the rapid progress made by electrotechnics, it has thus already become possible to evolve from the famous old discovery of the chemical combination of nitrogen and oxygen in an electric spark, a technical method for the production of saltpetre.

The nitrate problem, as we know, has gained, within a short space of time, a very important place among technical problems, a fact which is due to the enormous consumption of Chili saltpetre.

Commercial activities

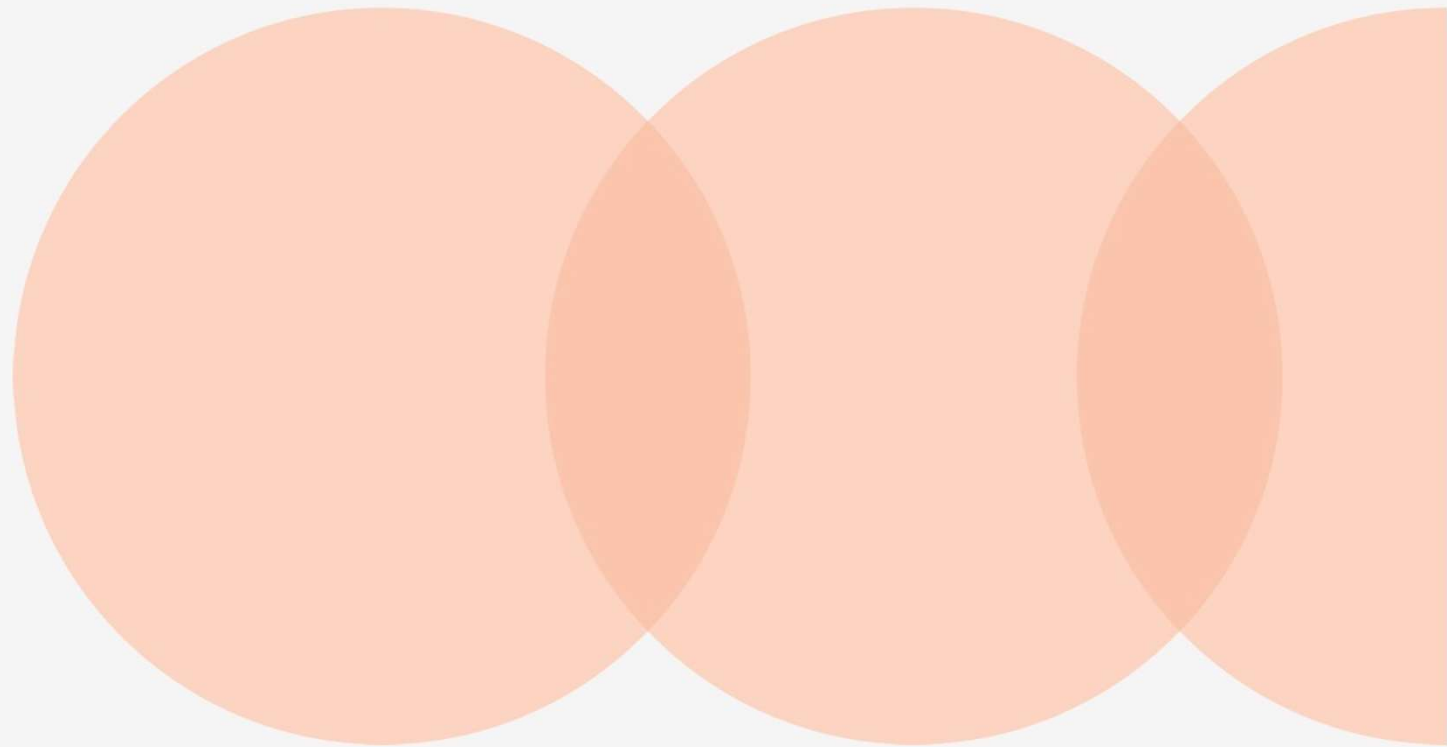
Vitalfluid



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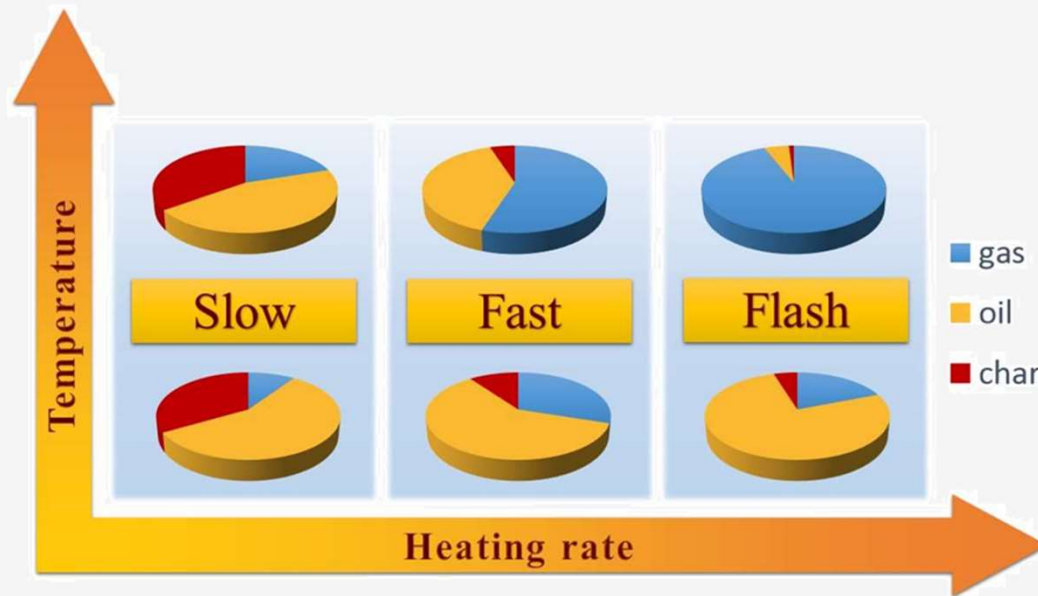
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Plastic waste



Plastic recycling

Gasification: conversion of plastic containing waste with oxygen to syngas (CO and H₂) -to chemicals



SGH2 Energy builds world's largest waste-to-hydrogen plant near L.A.

The world's largest green hydrogen plant is being developed by SGH2 Energy in Lancaster, a city situated some 70 miles north of Los Angeles. The facility will transform plastic waste into hydrogen for use as a fuel for transport and power generation.



CO₂ utilization First results in the 1978

The physics of a chemically active plasma with nonequilibrium vibrational excitation of molecules

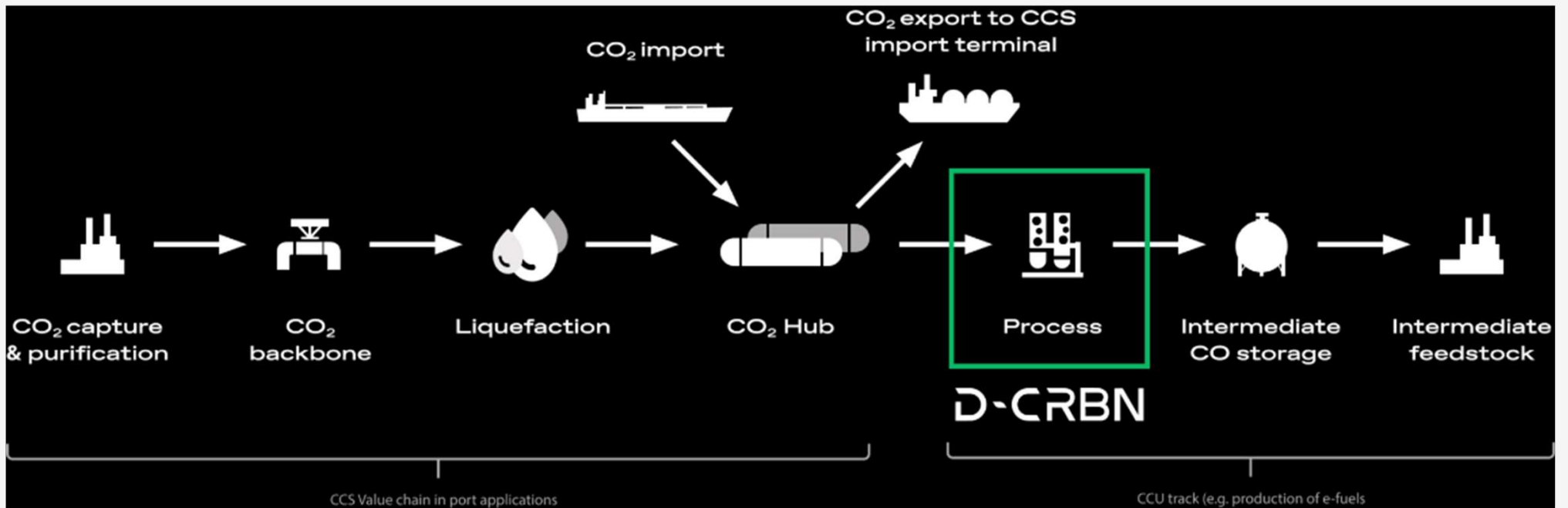
V. D. Rusanov, A. A. Fridman, and G. V. Sholin

I. V. Kurchatov Institute of Atomic Energy
Usp. Fiz. Nauk **134**, 185–235 (June 1981)

The results of experimental and theoretical studies of chemical reactions in nonequilibrium plasmas are reviewed. Special attention is given to processes stimulated by vibrational excitation of the ground electronic state of the reacting molecules in the plasma. General patterns in the kinetics of these reactions are discussed, and the optimum discharge parameters for maximum energy efficiency are noted. Specific plasma-chemical processes—the dissociation of CO₂ and H₂O and the synthesis of nitrogen oxides—are described. Experimental results are presented for hf, microwave, glow, plasma-beam, and non-self-sustained discharges, for plasma radiolysis, etc.

PACS numbers: 52.40. — w, 52.80. — s

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Plasma technology at Brightsite

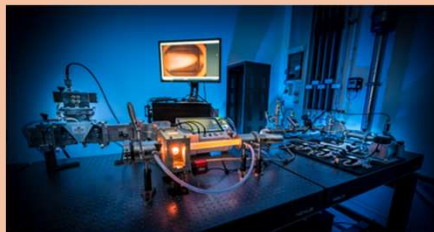
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Three generations in parallel



Lab Scale
TRL 1-3

- Started in 2020
- R&D program for the direct formation of Ethylene using Microwave plasma (1-10 kW)

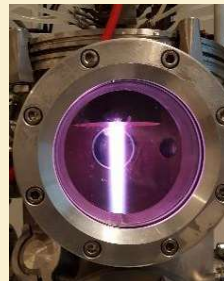


Plasma Lab at Brightsite



Bench Scale
TRL 4-5

- Started in 2021/2022
- R&D program to optimize the Hüls process using arc technology and microwave plasma (50 kW)



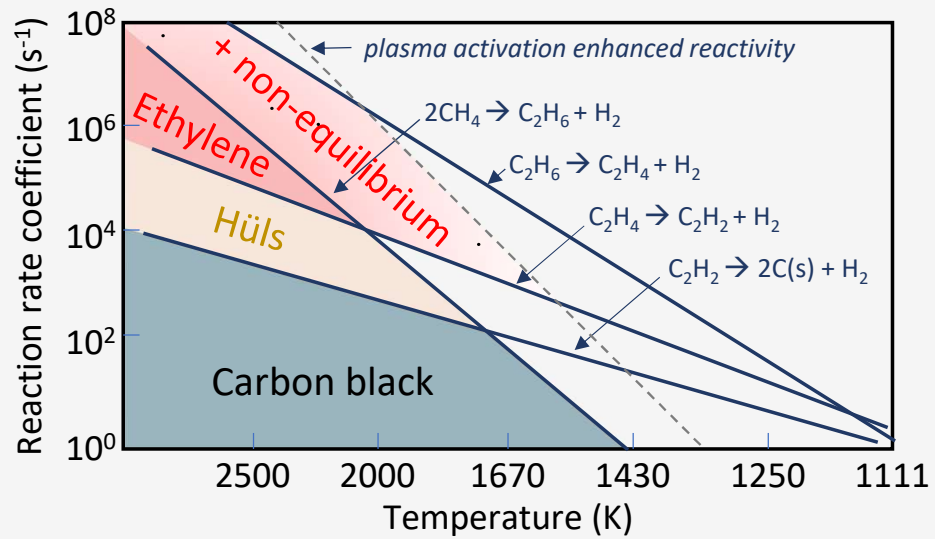
Pilot Plant
TRL 6-7

- Started in 2022/2023
- Hüls process for the formation of hydrogen and acetylene using arc technology (150~500 kW)

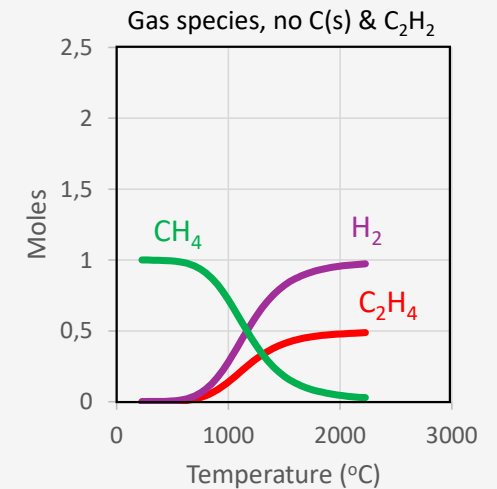
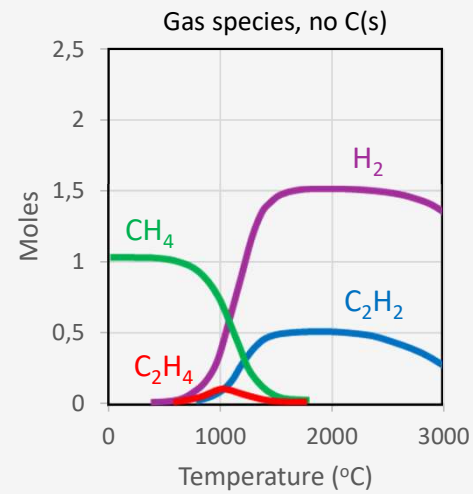
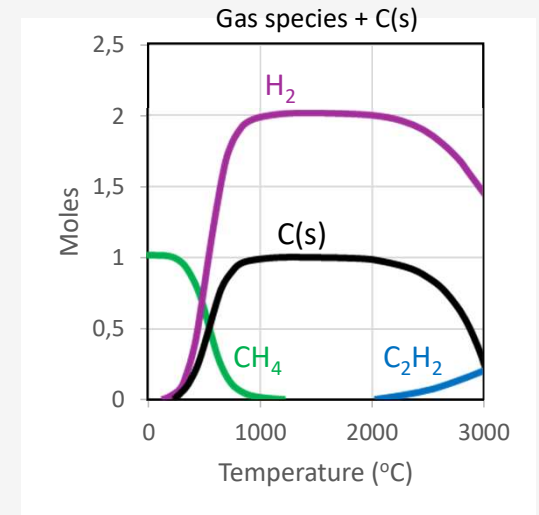


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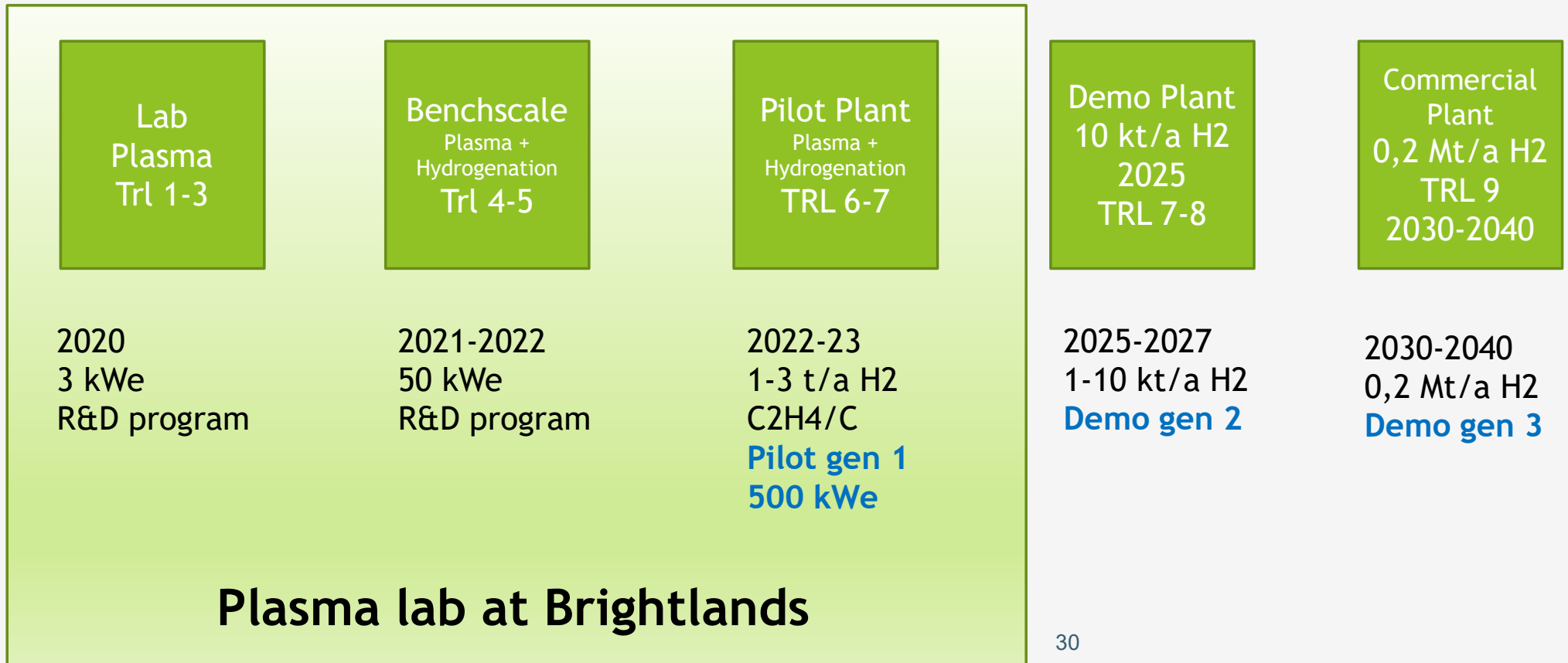
G van Rooij



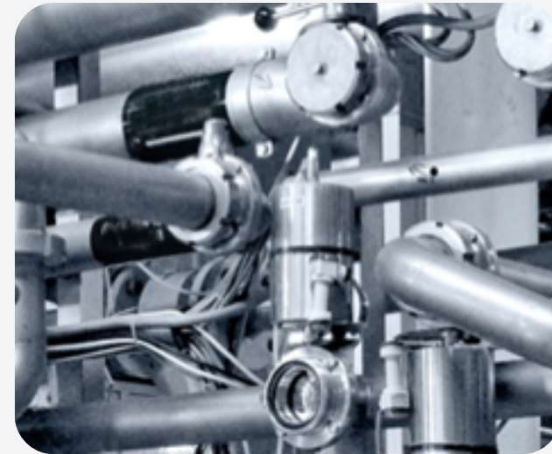
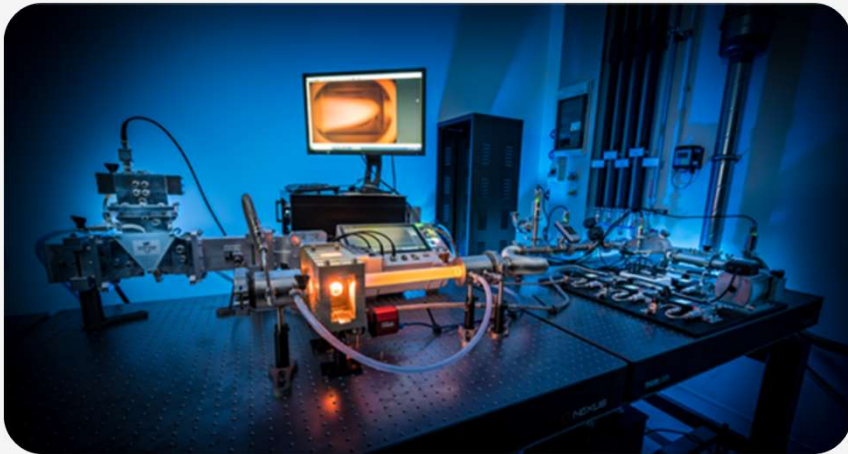
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Timeline plasma technology

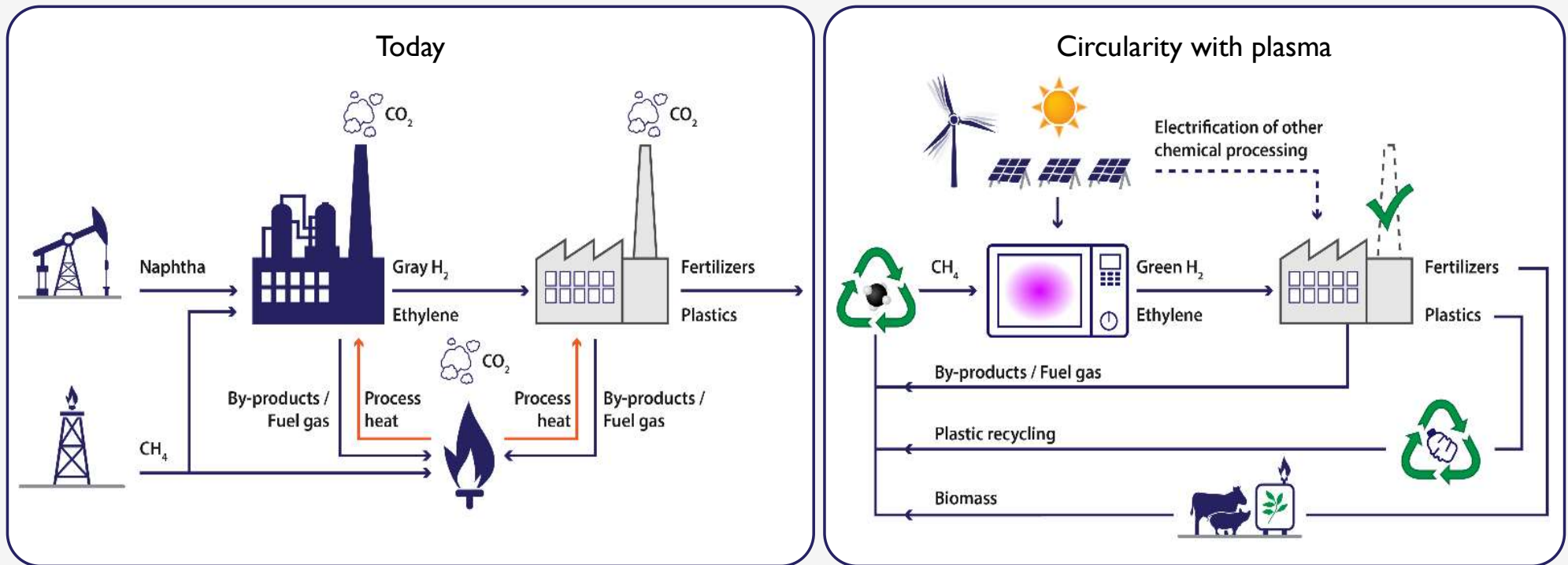
2020-2040



Accelerate scale-up by collaborative innovation



Plasma technology an important step towards the ultimate circular chemistry



Courtesy Gerard van Rooij

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